EXPERIMENTAL AND NUMERICAL INVESTIGATION OF FLAMELESS COMBUSTION OF LOW CALORIFIC VALUE GASES FOR THE IRON AND STEEL INDUSTRY

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Within the iron and steel industry, waste gases such as blast furnace gas (BFG) and Coal Oven Gas (COG) are produced and can be used to fuel gas turbines and generate power. These gases are mainly composed of CH4, CO and H2 in variable proportions. While generation of a stable flame is difficult because of the variable calorific value and ignition characteristics of these gases, diluted flameless combustion can simultaneously be stable and produce low emissions [1] [2].

Flameless, or diluted, combustion is achieved by preheating the combustion air and fuel above the auto-ignition temperature of the mixture and by creating a strong recirculation of burnt gases inside the combustion chamber, which leads to a dilution of the fuel-oxidizer mixture. This causes the combustion region to be extended in the whole furnace, rather than concentrated on a flame front as in common burners. Dilution avoids the formation of thermal hot spots, significantly reducing nitrogen oxides and carbon emissions without compromising the efficiency [3] [4].

The aim of this work is to characterize the combustion features of low calorific fuels (blends of CO, CH4, H2 and N2) in diluted combustion conditions, with focus on the effects of fuel composition on thermal efficiency and pollutants production. This study is based on both numerical simulations and experiments on a 30 kW combustion chamber. Combustion regimes [5] and properties are characterized by means of measurements of gas composition inside the combustion chamber, flue gas temperature and composition (including NOx) and intensity of the chemiluminescence emission of OH radicals in the reaction zone.

References

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