

INFLUENCE OF HYDROGEN ADDITION TO METHANE UNDER MILD CONDITIONS

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Abstract

Environmental issues related to CO₂ emissions deriving from power and energy sectors are amongst the most relevant industrial challenges [1]. In particular, an effective approach to address the needed CO₂ emissions reduction is based on the gradual replacement of hydrocarbon-based fuels with hydrogen. However, many techno-economic barriers hinder a straightforward transition to a hydrogen economy [2]. The use of hydrocarbon/hydrogen blends has been proposed as intermediate and bridge solution, thus providing a transition strategy to a carbon-free energy system in the future. In addition, MILD combustion [3] was already proven as suitable and effective technology. However, the step toward the technological readiness of MILD based devices has not yet been completed. In this framework, the present study wants to shed light on the combustion features of methane/hydrogen blends and on the related pollutant emissions in a truly MILD burner. The influence of reactive mixture composition (ϕ) and the hydrogen content in the fuel mixture (%H₂ vol) was highlighted, focusing on oxidation process stability and NO_x emissions. Results confirm the wide fuel flexibility of the MILD combustion process, whose performance is totally released from the fuel mixtures combustion properties. Specifically, results testified the ability of efficiently convert CH₄/H₂ blends by substantially keeping unaffected operating temperatures and NO_x emissions to the levels typically detected for pure methane combustion. In this respect, kinetic analyses highlighted the main reaction involved in NO_x production/consumption for CH₄/H₂ blends and the key role of radical species deriving from both the fuels. In particular, the increased radical pool concentration deriving from the increasing %H₂, along with the resulting decreasing ones from CH₄, balance each other, thus not affecting the NO_x net production.

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